

## Concise Syntheses of the Novel 1*H*-Pyrrolo[3,2-*g*]quinazoline Ring System and its [2,3-*f*] Angular Isomer

H. D. Hollis Showalter,<sup>\*,†</sup> Li Sun,<sup>†</sup> Anthony D. Sercel,<sup>†</sup>  
R. Thomas Winters,<sup>†</sup> William A. Denny,<sup>‡</sup> and  
Brian D. Palmer<sup>‡</sup>

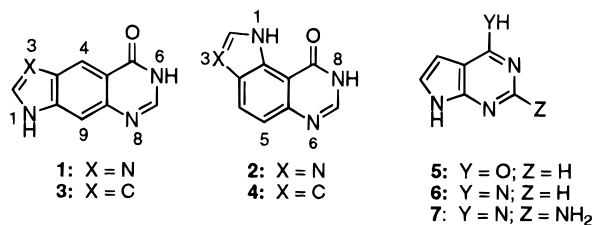
Chemistry Department, Parke-Davis Pharmaceutical  
Research, Division of Warner-Lambert Company, 2800  
Plymouth Road, Ann Arbor, Michigan 48106-1047 and  
Cancer Research Laboratory, University of Auckland School  
of Medicine, Private Bag 92019, Auckland, New Zealand

Received October 2, 1995

### Introduction

Several years ago, Leonard and co-workers reported on the synthesis of a series of linear benzo-substituted purines, agents that act as purine mimics in many enzymatic reactions.<sup>1</sup> These are chemically designated as imidazo[4,5-*g*]quinazolines and are exemplified by *lin*-benzohypoxanthine (**1**). There soon followed the synthesis of a number of angular analogues<sup>2</sup> including the imidazo[4,5-*f*] series of compounds exemplified by *prox*-benzohypoxanthine (**2**). More recently, others have extended this work to the synthesis of isomeric forms of these tricycles including linear<sup>3</sup> and angular<sup>4</sup> pyrazoloquinazoline systems. However, reports describing the synthesis of corresponding benzo-substituted deazapurines (*i.e.*, pyrroloquinazolines) are sparse. Apart from the synthesis of a single angular isomer (pyrrolo[3,2-*f*]quinazoline)<sup>5</sup> of this heterocyclic system, these ring systems have never been made. We report in this paper approaches toward the first synthesis of a linear pyrroloquinazoline and one of its angular isomers, exemplified by compounds **3** and **4**, respectively. Heterocycle **3** can be viewed as a linear extension of the deazapurine **5**<sup>6</sup> (*i.e.*, pyrrolo[2,3-*d*]pyrimidine ring system), which is an important precursor of the 4-amino analogue **6**.<sup>7</sup> Heterocycle **6** and its 2-amino derivative **7** are structural elements of the important antitumor nucleoside tuber-

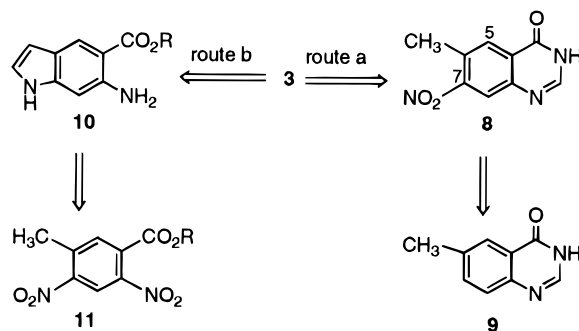
cidin,<sup>8</sup> which is highly stable toward acid hydrolysis, and a series of novel antifolate inhibitors of dihydrofolate reductases,<sup>9</sup> respectively.



### Results and Discussion

To construct our targeted 1*H*-pyrrolo[3,2-*g*]quinazoline ring system, exemplified by **3**, we initially adopted a strategy whereby the indole ring would be annulated via Batcho–Leimgruber indole synthesis<sup>10</sup> onto a 7-nitroquinazolinone precursor **8** (Scheme 1, route a). Accordingly, nitration of 6-methylquinazolin-4-one<sup>11</sup> (**9**) was evaluated under a number of conditions. However, this provided predominately the 5-nitro isomer<sup>3c,4d,e</sup> instead of the desired 7-isomer **8**. Thus, we pursued the construction of **3** by a reverse sequence in which pyrimidine ring annulation was carried out on a suitably functionalized indole precursor **10** (Scheme 1, route b).

#### Scheme 1



The synthesis of linear target **3** is shown in Scheme 2. Fischer esterification of commercially available 3-methyl-4-nitrobenzoic acid (**12**) gave ethyl ester **13**<sup>12</sup> in 92% yield, which was then nitrated regioselectively with cold fuming nitric acid to afford an 86:14 mixture of desired product **14**<sup>13</sup> and its isomer **15** in 98% yield. Ester **14** was purified to >97% isomeric purity in an overall yield of 70% by crystallization from 2-propanol. At this juncture, **14** could be parlayed into target compound **3** by one of two routes. In the first, Batcho–Leimgruber indole synthesis was performed by condensation of **14** with DMF dimethyl acetal in refluxing *p*-dioxane to cleanly give the enamino ester **16**, which was not isolated but was directly reductively cyclized via catalytic hydrogenation to the indolic anthranilate **17** in 96% yield. Indole **17** was contaminated with 4–10% of the corresponding 1-hy-

(8) Anderson, J. D.; Bontems, R. J.; Geary, S.; Cottam, H. B.; Larson, S. B.; Matsumoto, S. S.; Smee, D. F.; Robins, R. K. *Nucleosides Nucleotides* **1989**, *8*, 1201–1216.

(9) Gangjee, A.; Mavandadi, F.; Queener, S. F.; McGuire, J. J. *J. Med. Chem.* **1995**, *38*, 2158–2165.

(10) Clark, R. D.; Repke, D. B. *Heterocycles* **1984**, *22*, 195–221.

(11) Stevenson, T. M.; Kazmierczak, F.; Leonard, N. J. *J. Org. Chem.* **1986**, *51*, 616–21.

(12) Beilstein, F.; Kreisler, U. *Justus Liebigs Ann. Chem.* **1867**, *144*, 163–184.

(13) Blatt, A. H. *J. Org. Chem.* **1960**, *25*, 2030–2034.

<sup>†</sup> Parke-Davis Pharmaceutical Research.

<sup>‡</sup> University of Auckland School of Medicine.

(1) (a) Leonard, N. J.; Morrice, A. G.; Sprecker, M. A. *J. Org. Chem.* **1975**, *40*, 356–363. (b) Moder, K. P.; Leonard, N. J. *J. Am. Chem. Soc.* **1982**, *104*, 2613–2624. (c) Leonard, N. J.; Kazmierczak, F.; Rykowski, A. Z. *J. Org. Chem.* **1987**, *52*, 2933–2935. (d) Leonard, N. J.; Petric, A.; Rykowski, A. *J. Org. Chem.* **1988**, *53*, 3873–3875.

(2) Imidazo[4,5-*f*]quinazolines: (a) Morrice, A. G.; Sprecker, M. A.; Leonard, N. J. *J. Org. Chem.* **1975**, *40*, 363–366. (b) Schneller, S. W.; Ibay, A. C. *J. Org. Chem.* **1986**, *51*, 4067–4070. Imidazo[4,5-*h*]quinazolines: (c) ref 2a.

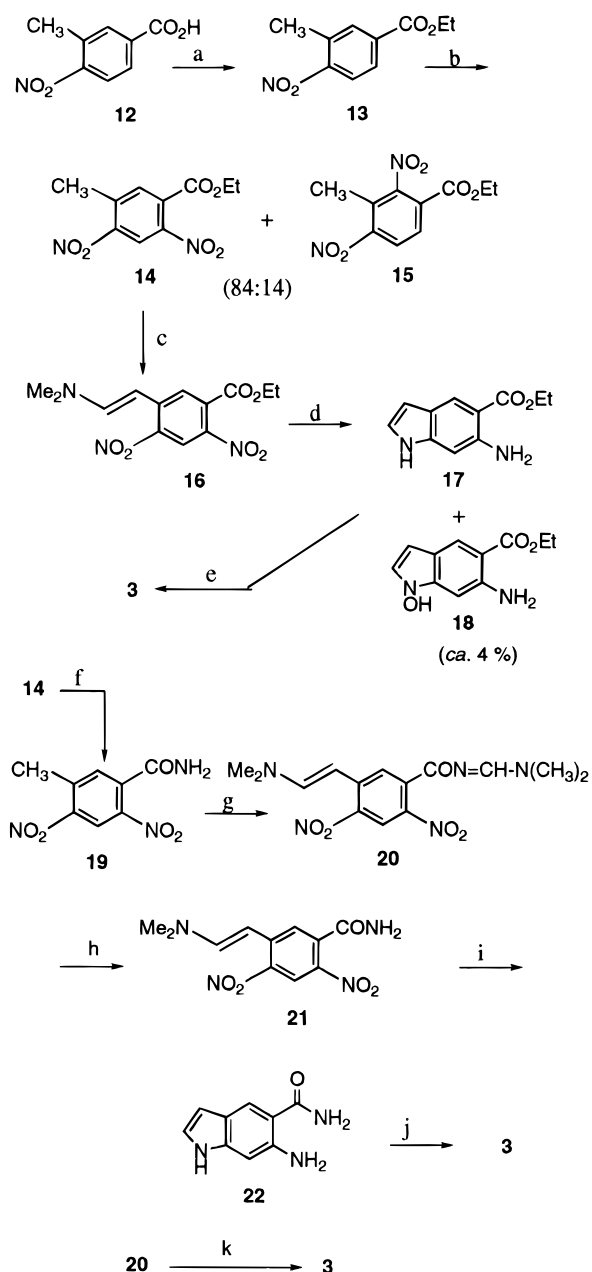
(3) Pyrazolo[3,4-*g*]quinazolines: (a) Lichtenthaler, F. W.; Moser, A. *Tetrahedron Lett.* **1981**, *22*, 4397–4400. Pyrazolo[4,3-*g*]quinazolines: (b) Foster, R. H.; Leonard, N. J. *J. Org. Chem.* **1979**, *44*, 4609–4611. (c) Cuny, E.; Lichtenthaler, F. W.; Moser, A. *Tetrahedron Lett.* **1980**, *21*, 3029–3032. (d) Cuny, E.; Lichtenthaler, F. W.; Jahn, U. *Chem. Ber.* **1981**, *114*, 1624–1635.

(4) Pyrazolo[3,4-*h*]quinazolines: (a) ref 3a; Pyrazolo[4,3-*f*]quinazolines: (b) ref 3d; Pyrazolo[3,4-*f*]quinazolines: (c) ref 3c. (d) Foster, R. H.; Leonard, N. J. *J. Org. Chem.* **1980**, *45*, 3072–3077. (e) Lichtenthaler, F. W.; Cuny, E. *Heterocycles* **1981**, *15*, 1053–1059.

(5) (a) Jones, M. L.; Kuyper, L. F.; Styles, V. L.; Caddell, J. M. *J. Heterocycl. Chem.* **1994**, *31*, 1681–1683. (b) Kuyper, L. F.; Jones, Michael L.; Baccanari, D. P. European Patent Application 542497; *Chem. Abstr.* **1994**, *120*, 8606u.

(6) Seela, F.; Richter, R. *Chem. Ber.* **1978**, *111*, 2925–2930.

(7) Seela, F.; Menkhoff, S.; Behrendt, S. *J. Chem. Soc., Perkin Trans. 2* **1986**, 525–530.

Scheme 2<sup>a</sup>

<sup>a</sup> Key: (a) EtOH, H<sub>2</sub>SO<sub>4</sub>, reflux, 24 h, 92%; (b) fuming HNO<sub>3</sub>, H<sub>2</sub>SO<sub>4</sub>, -20 °C, 5.75 d, 70%; (c) DMFDMA, *p*-dioxane, reflux, 1.5 h; (d) 5% Pd/C, H<sub>2</sub> (60 psi), MeOH, 1 h, 96% two steps; (e) formamidine acetate, 2-methoxyethanol, reflux, 2 h, 64%; (f) anhyd NH<sub>3</sub>/MeOH, 80 °C, 12 h, 90%; (g) BBDM, DMF, 25 °C, 1 h, 84%; (h) MeOH, glacial HOAc, reflux, 2 d, 80%; (i) 10% Pd/C, H<sub>2</sub>, THF, 12 h, 72%; (j) (EtO)<sub>3</sub>CH, 100 °C, 1 h, 50%; (k) RaNi, H<sub>2</sub> (1500 psi), 1:1 THF:MeOH, 25 °C, 57%.

droxyindole **18** that displayed the same *R<sub>f</sub>* by TLC.<sup>14</sup> Annulation of the pyrimidinone ring onto **17** proceeded smoothly in 64% yield with formamidine acetate<sup>15</sup> in

(14) Although **18** is not observed in the EIMS of the mixture, it is evident by high-field NMR. Thus, in the Me<sub>2</sub>SO-*d*<sub>6</sub> <sup>1</sup>H NMR spectrum of the mixture of **17** and **18**, H-2 and H-3 of the indole ring of **17** resonate at δ 7.09 and 6.29, respectively, and display couplings to the indolic NH proton as well as to each other. Impurity **18** is evidenced by the presence of the additional resonances at δ 7.30 and 6.23, which are coupled only to each other, and another exchangeable resonance at δ 5.87. In the CDCl<sub>3</sub> <sup>13</sup>C NMR spectrum, there are additional resonances at δ 132.96, 122.93, 100.02, 59.64 and 16.26. The 132.96 and 100.02 signals are the most prominent, and correspond to CH carbons from the DEPT spectrum. Therefore, they most likely correspond to C-2 and C-3, respectively, of **18**.

refluxing 2-methoxyethanol to provide target compound **3** without any *N*-hydroxy impurity. More classical methods of ring closure including the Niementowski reaction (hot formamide)<sup>16</sup> and s-triazine in refluxing alcohol<sup>17</sup> failed.

In a second route to **3**, shown in Scheme 2, aminolysis of dinitro ester **14** to amide **19**<sup>18</sup> followed by condensation with *tert*-butoxybis(dimethylamino)methane (Bredereck's reagent) in DMF at room temperature afforded the enamino amidine **20** as a brick red solid in 76% yield over two steps.<sup>19</sup> Similar reaction with DMF dimethyl acetal proceeded much less cleanly. Since **20** contains the necessary framework for further elaboration to target **3** via tandem ring closure, we subjected it to various conditions of reductive cyclization, including numerous conditions of catalytic hydrogenation over 10% Pd/C, and selected dissolving metal reductions.<sup>20</sup> Amongst these, the best condition developed involved high pressure Raney nickel hydrogenation in 1:1 THF:MeOH<sup>21</sup> at high dilution to afford compound **3** in 57% yield on a small scale. However, larger scale reactions were highly irreproducible, with 10–15% yields the norm. Thus, we found it more convenient to selectively hydrolyze the amidine side chain of **20** with mildly acidic methanol to give the enamino amide **21** in 80% yield. Reductive ring closure similar to that described above for **16** gave the indole amide **22** in 72% yield.<sup>22</sup> Compound **22** as well as **17** are quite air sensitive and darken while in solution. Ring closure of **22** to target **3** proceeded uneventfully in hot triethyl orthoformate. Solutions of **3** in common organic solvents are also unstable, but it can be stored as a solid without significant decomposition.

The synthesis of angular target compound **4**, shown in Scheme 3, proceeded in a much more straightforward fashion. Thus, Fischer esterification of 3-methyl-6-nitrobenzoic acid **23** provided ester **24** in 91% yield. Nitration of **24** with fuming nitric acid in triflic acid<sup>23</sup> proceeded to give a 1:1 mixture of compound **25**<sup>13</sup> and its regioisomer **14** in 61% yield. Following separation by careful column chromatography, bis-nitro ester **25** was condensed with DMF dimethyl acetal in DMF at 100 °C

(15) Albert, A. In *Advances in Heterocyclic Chemistry*; Katritzky, A. R., Ed.; Academic: New York, 1982; Vol 32, pp 1–81.

(16) Irwin, W. J.; Wibberley, D. G. In *Advances in Heterocyclic Chemistry*; Katritzky, A. R., Boulton, A. J., Eds.; Academic: New York, 1969; Vol 10, pp 149–198.

(17) Kreuztberger, A.; Uzbek, M. U. *Arch. Pharm. (Weinheim)* **1972**, *305*, 171–175.

(18) Piskov, V. B.; Osanova, L. K.; Koblova, I. A. *Zh. Org. Khim.* **1970**, *6*, 559–564; *Chem. Abstr.* **1970**, *72*, 132251d.

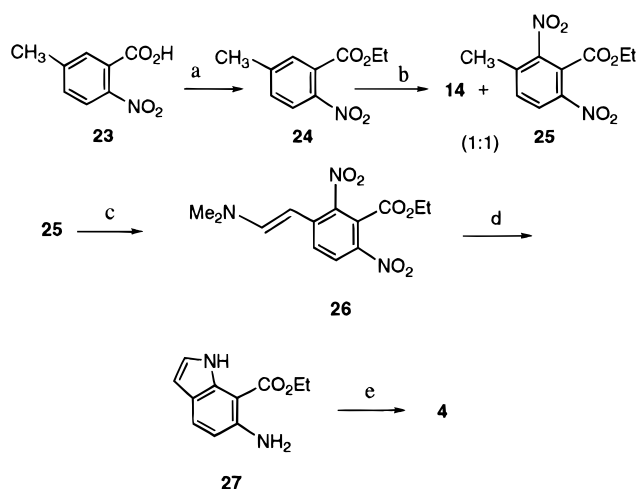
(19) It is imperative to use pure (*tert*-butoxy)bis(dimethylamino)methane in this reaction. In one run utilizing old reagent, the reaction did not proceed cleanly. Attempted purification of crude **20** by column chromatography resulted in significant decomposition, in part to amide **21**.

(20) (a) Aqueous TiCl<sub>3</sub>: Lloyd, D. H.; Nichols, D. E. *Tetrahedron Lett.* **1983**, *24*, 4561–4562. (b) Fe/HOAc/EtOH: Ponticello, G. S.; Baldwin, J. J. *J. Org. Chem.* **1979**, *44*, 4003–4005. (c) Zn/HOAc: Glushkov, R. G.; Volskova, V. A.; Magidson, O. Y. *Khim.-Farm. Zh.* **1967**, *1*, 25–32; *Chem. Abstr.* **1968**, *68*, 105143f.

(21) Hengartner, U.; Batcho, A. D.; Blount, J. F.; Leimgruber, W.; Larscheid, M. E.; Scott, J. W. *J. Org. Chem.* **1979**, *44*, 3748–3752.

(22) In a separate run in which the reaction was worked up before there was theoretical uptake of H<sub>2</sub>, we isolated a component, mp 171 °C dec, ca. 87% pure by <sup>1</sup>H NMR, that displayed the same *R<sub>f</sub>* by TLC as desired product **22**. Its <sup>1</sup>H NMR and CIMS suggest that it is the intermediate 1-hydroxy analogue of **22**, corresponding to **18** observed in the reduction **16** to **17**:<sup>14</sup> <sup>1</sup>H NMR (Me<sub>2</sub>SO-*d*<sub>6</sub>) δ 10.92 (s, exchanges with D<sub>2</sub>O, 1H), 7.74 (s, 1H), 7.65 (br s, exchanges with D<sub>2</sub>O, 1H), 7.15 (d, *J* = 3.4 Hz, 1H), 6.94 (br s, exchanges with D<sub>2</sub>O, 1H), 6.55 (s, 1H), 6.22 (s, exchanges with D<sub>2</sub>O, 2H), 6.11 (d, *J* = 3.4 Hz, 1H); CIMS *m/z* (relative intensity) 192 (14), 191(M<sup>+</sup>, 40), 175 (100), 174 (43).

(23) Coon, C. L.; Hill, M. E. U.S. Patent 3714272; *Chem. Abstr.* **1973**, *78*, 97303x.

Scheme 3<sup>a</sup>

<sup>a</sup> Key: (a) EtOH, H<sub>2</sub>SO<sub>4</sub>, reflux, 2 d, 91%; (b) fuming HNO<sub>3</sub>, CF<sub>3</sub>SO<sub>3</sub>H, CH<sub>2</sub>Cl<sub>2</sub>, 25 °C, 20 h, 31%; (c) DMFDMA, DMF, 100 °C, 4 h, 70%; (d) 10% Pd/C, H<sub>2</sub> (50 psi), MeOH, 4 h, 48% (e) formamide, 150–180 °C, 3 h, 44%.

to afford compound **26** in 70% yield. Reductive cyclization of **26** via catalytic hydrogenation to indole **27** followed by condensation with formamide at 100 °C gave target pyrrolo[2,3-*f*]quinazolinone **4** in 21% yield over two steps. Yields for the synthesis of angular target **4** were not optimized.

In summary, we have developed synthetic pathways to two novel pyrroloquinazolinone heterocycles, benzo-substituted variants of the biologically important pyrrolo[2,3-*d*]pyrimidine ring system. We expect these heterocycles to find broad application as novel templates in organic synthesis and drug design.

### Experimental Section

Melting points are uncorrected. Column chromatography was carried out in the flash mode utilizing E. Merck 230–400-mesh silica gel. Analytical TLC was carried out on E. Merck silica gel 60 F<sub>254</sub> plates with detection by UV light. Palladium on activated carbon utilized in catalytic hydrogenations was purchased from Aldrich Chemical Co. All reaction solvents were reagent grade or distilled-in-glass and were stored over activated 3A (for lower alcohols) or 4A molecular sieves. Following normal workup procedures, organic extracts were dried over anhydrous Na<sub>2</sub>SO<sub>4</sub> or MgSO<sub>4</sub> prior to concentration.

**3-Methyl-4-nitrobenzoic Acid, Ethyl Ester (13).** A solution of 102.1 g (564 mmol) of 3-methyl-4-nitrobenzoic acid (**12**) dissolved in 1 L of absolute EtOH was carefully treated with 10 mL of concd H<sub>2</sub>SO<sub>4</sub>. The solution was heated at reflux for 24 h then concentrated to 300 mL by allowing the EtOH to distill off. Upon cooling solids precipitated. The suspension was further cooled to about –10 °C, and then the solids were collected, washed with a small amount of cold EtOH, and recrystallized from 450 mL of boiling hexanes to afford 108 g (92%) of **13** as a pale yellow solid: mp 52.5–53.5 °C (lit.<sup>12</sup> mp 55 °C); <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 11.53 (s, 1H), 7.98–7.93 (m, 3H), 4.38 (q, *J* = 7.0 Hz, 2H), 2.58 (s, 3H), 1.38 (t, *J* = 7.0 Hz, 3H). Anal. Calcd for C<sub>10</sub>H<sub>11</sub>NO<sub>4</sub>: C, 57.41; H, 5.30; N, 6.70. Found: C, 57.23; H, 5.25; N, 6.65.

**4,6-Dinitro-*m*-toluic Acid, Ethyl Ester (14).** A solution of 112 g (535 mmol) of ester **13** in 460 mL of concd H<sub>2</sub>SO<sub>4</sub> was cooled to –10 °C and then treated dropwise over a 45 min period with 106 mL of fuming HNO<sub>3</sub> so as to maintain the internal temperature at –10 °C. The solution was stored at –20 °C for 5.75 d. The resulting golden yellow syrup was carefully poured into about 3.2 L of crushed ice precipitating out a waxlike solid. The suspension was extracted into EtOAc (3 × 600 mL), and the combined extracts were washed with 5% aq NaOH until the aqueous phase remained neutral and then dried. Concentration

gave 133 g of an oil that solidified as an 86:14 mixture of isomers (monitored by <sup>1</sup>H NMR or TLC utilizing 4:1 hexanes:EtOAc). The crude solid was triturated in 150 mL of *i*-PrOH and then filtered to leave a pale white solid. The solid was dissolved in 400 mL of hot *i*-PrOH, and the solution was allowed to slowly cool to 25 °C. After storage the solid was collected, washed with *i*-PrOH, and dried to leave 95 g (70%) of **14**, 98% by <sup>1</sup>H NMR, as a white solid: mp 60.5–62.5 °C (lit.<sup>13</sup> mp 47–48 °C); <sup>1</sup>H NMR (Me<sub>2</sub>SO-*d*<sub>6</sub>) δ 8.61 (s, 1H), 8.00 (s, 1H), 4.32 (q, *J* = 7.1 Hz, 2H), 2.63 (s, 3H), 1.24 (t, *J* = 7.1 Hz, 3H); <sup>13</sup>C NMR (Me<sub>2</sub>SO-*d*<sub>6</sub>) δ 164.1, 149.8, 145.4, 140.2, 134.4, 130.7, 121.3, 63.2, 19.8, 14.0; CIMS *m/z* (relative intensity) 255 (MH<sup>+</sup>, 36). Anal. Calcd for C<sub>10</sub>H<sub>10</sub>N<sub>2</sub>O<sub>6</sub>: C, 47.25; H, 3.97; N, 11.02. Found: C, 47.53; H, 4.05; N, 11.00.

The filtrate was concentrated to a ca. 1:1 mixture of isomers. Dissolution of the solid in dilute 2:1 hexanes:*i*-PrOH followed by storage at 25 °C for several weeks resulted in the crystallization of >90% pure material. An additional crystallization from *i*-PrOH gave pure **15**: mp 75.5–76 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 8.04 (d, *J* = 8.7 Hz, 1H), 7.99 (d, *J* = 8.7 Hz, 1H), 4.41 (q, *J* = 7.0 Hz, 2H), 2.45 (s, 3H), 1.38 (t, *J* = 7.2 Hz, 3H); CIMS *m/z* (relative intensity) 255 (MH<sup>+</sup>, 36), 209 (100). Anal. Calcd for C<sub>10</sub>H<sub>10</sub>N<sub>2</sub>O<sub>6</sub>: C, 47.25; H, 3.97; N, 11.02. Found: C, 47.31; H, 3.97; N, 11.03.

**6-Amino-1*H*-indole-5-carboxylic Acid, Ethyl Ester (17).** A solution of 15.0 g (59 mmol) of ester **14** and 9.4 mL (71 mmol) of DMF dimethyl acetal in 250 mL of *p*-dioxane was heated at reflux for 1.5 h. The resulting deep red solution, containing (*E*)-5-[2-(dimethylamino)ethenyl]-2,4-dinitrobenzoic acid, ethyl ester (**16**) as the sole product, was cooled to 25 °C, diluted with 100 mL of MeOH and then hydrogenated at 60 psi over 8.0 g of 5% Pd/C. After 1 h, when H<sub>2</sub> uptake had ceased, the mixture was filtered and concentrated to give a solid that was purified by column chromatography. Elution with EtOAc/petroleum ether (1:1) afforded 11.6 g (96%) of **17**, contaminated with ca. 4% of the corresponding 1-hydroxyindole **18**.<sup>14</sup> Crystallization from aq MeOH provided pure **17**: mp 139 °C; <sup>1</sup>H NMR (Me<sub>2</sub>SO-*d*<sub>6</sub>) δ 10.68 (br s, exchanges with D<sub>2</sub>O, 1H), 8.02 (s, 1H), 7.09 (dd, *J* = 3.1, 2.4 Hz, 1H), 6.64 (s, 1H), 6.29 (m, 1H), 6.18 (br s, exchanges with D<sub>2</sub>O, 2H), 4.24 (q, *J* = 7.0 Hz, 2H), 1.32 (t, *J* = 7.0 Hz, 3H); <sup>13</sup>C NMR (CDCl<sub>3</sub>) δ 168.8, 146.4, 140.5, 124.8, 123.9, 120.6, 107.9, 103.5, 96.0, 60.1, 14.5; EIMS *m/z* (relative intensity) 204 (M<sup>+</sup>, 92), 158 (100), 130 (59), 104 (19). Anal. Calcd for C<sub>11</sub>H<sub>12</sub>N<sub>2</sub>O<sub>6</sub>: C, 64.69; H, 5.92; N, 13.72. Found: C, 64.39; H, 6.04; N, 13.44.

**5-Methyl-2,4-dinitrobenzamide (19).** A solution of 5.0 g (24 mmol) of ester **14** in 50 mL of saturated anhydrous NH<sub>3</sub> in MeOH was stirred in a pressure reactor at 80 °C for 12 h. The mixture was concentrated, and the resulting black residue was purified by column chromatography eluting with EtOAc:hexanes (1:1) to give 4.0 g (90%) of **19**: mp 173–174 °C; <sup>1</sup>H NMR (Me<sub>2</sub>SO-*d*<sub>6</sub>) δ 8.63 (s, 1H), 8.25 (s, exchanges D<sub>2</sub>O, 1H), 7.93 (s, exchanges D<sub>2</sub>O, 1H), 7.81 (s, 1H), and 2.63 (s, 3H); <sup>13</sup>C NMR (Me<sub>2</sub>SO-*d*<sub>6</sub>) δ 165.7, 148.1, 144.3, 139.1, 136.0, 133.2, 120.6, 19.3; CIMS *m/z* (relative intensity) 226 (MH<sup>+</sup>, 38), 209 (100). Anal. Calcd for C<sub>8</sub>H<sub>7</sub>N<sub>3</sub>O<sub>5</sub>: C, 42.68; H, 3.13; N, 18.66. Found: C, 42.74; H, 3.15; N, 18.43.

***N*-[(Dimethylamino)methylene]-(*E*)-5-[2-(dimethylamino)ethenyl]-2,4-dinitrobenzamide (20).** A solution of 2.25 g (10 mmol) of benzamide **19** in 10 mL of DMF was treated with 6.2 mL (30 mmol) of (*tert*-butyloxy)bis(dimethylamino)methane. The reaction mixture was stirred at 25 °C for 1 h. The solvent was evaporated *in vacuo*, and the residue was suspended in H<sub>2</sub>O. The precipitate was collected then washed successively with H<sub>2</sub>O and Et<sub>2</sub>O to afford 2.76 g (84%) of **20** as a red solid: mp 218–219 °C; <sup>1</sup>H NMR (Me<sub>2</sub>SO-*d*<sub>6</sub>) δ 8.55 (s, 1H), 8.47 (s, 1H), 8.04 (d, *J* = 13.0 Hz, 1H), 7.76 (1H, s), 5.95 (d, *J* = 13.0 Hz, 1H), 3.21 (s, 3H), 3.00 (m, 9H); <sup>13</sup>C NMR (Me<sub>2</sub>SO-*d*<sub>6</sub>) 176.0, 160.5, 151.5, 141.6, 139.0, 138.9, 137.1, 123.3, 122.1, 88.9, 40.9, 35.0; CIMS *m/z* (relative intensity) 336 (MH<sup>+</sup>, 43), 335 (12). Anal. Calcd for C<sub>14</sub>H<sub>17</sub>N<sub>5</sub>O<sub>5</sub>: C, 50.15; H, 5.11; N, 20.89. Found: C, 50.16; H, 4.94; N 20.59.

**(*E*)-5-[2-(Dimethylamino)ethenyl]-2,4-dinitrobenzamide (21).** A suspension of 13.0 g (38.8 mmol) of benzamide **20**, 500 mL of MeOH, and 2.5 mL of glacial HOAc was heated at reflux for 2 d. The hot mixture was filtered, the filtrate was cooled slowly to 25 °C, and then the precipitated solid was collected. Two additional crops were collected by concentrating

each respective mother liquor to a solid followed by crystallization from MeOH. A total of 8.7 g (80%) of **21**: mp 254–255 °C, was collected: <sup>1</sup>H NMR (Me<sub>2</sub>SO-*d*<sub>6</sub>) δ 8.54 (s, 1H), 8.14 (d, *J* = 12.8 Hz, 1H), 7.99 (br s, exchanges with D<sub>2</sub>O, 1H), 7.76 (s, 1H), 7.71 (br s, exchanges with D<sub>2</sub>O, 1H), 5.93 (d, *J* = 13.0 Hz, 1H); <sup>13</sup>C NMR (Me<sub>2</sub>SO-*d*<sub>6</sub>) δ 167.7, 152.6, 142.4, 139.6, 137.0, 136.5, 124.1, 122.8, 89.4; CIMS *m/z* (relative intensity) 281 (MH<sup>+</sup>, 34), 280 (20), 264 (100). Anal. Calcd for C<sub>11</sub>H<sub>12</sub>N<sub>4</sub>O<sub>5</sub>: C, 47.15; H, 4.32; N, 19.99. Found: C, 47.11; H, 4.25; N, 19.62.

**6-Amino-1H-indole-5-carboxamide (22).** A mixture of 1.9 g (6.8 mmol) of enamine **21**, 0.96 g of 10% Pd/C, and 190 mL of THF was hydrogenated at 1 atm for 12 h. The mixture was filtered, and the filtrate was concentrated to a solid that was triturated in acetone/ether to afford 0.86 g (72%) of **22**, mp 234–235 °C dec, as an off-white solid. An analytical sample gave mp 239–240 °C dec: <sup>1</sup>H NMR (Me<sub>2</sub>SO-*d*<sub>6</sub>) δ 10.57 (s, exchanges with D<sub>2</sub>O, 1H), 7.76 (s, 1H), 7.67 (br s, exchanges with D<sub>2</sub>O, 1H), 7.06–7.03 (m, 1H; collapses to d, *J* = 3.1 Hz with D<sub>2</sub>O), 6.88 (br s, exchanges with D<sub>2</sub>O, 1H), 6.56 (s, 1H), 6.22 (d, *J* = 3.2 Hz, 1H), 6.07 (s, exchanges with D<sub>2</sub>O, 2H); CIMS *m/z* (relative intensity) 175 (M<sup>+</sup>, 58), 159 (100), 158 (71). Anal. Calcd for C<sub>9</sub>H<sub>9</sub>N<sub>3</sub>O: C, 61.70; H, 5.18; N, 23.99. Found: C, 61.52; H, 5.35; N, 23.59.

**1,6-Dihydro-5H-pyrrolo[3,2-*g*]quinazolin-5-one (3).** **Method A.** A solution of 10.3 g (50 mmol) of the aminoindole **17** and 6.3 g (60 mmol) of formamidinium acetate in 200 mL of 2-methoxyethanol was heated at reflux for 2 h and then concentrated to dryness. The residue was slurried in H<sub>2</sub>O and filtered to afford crude **3**. The solid was dissolved in excess 1 N aq NaOH, and the solution was filtered and then brought to neutrality with 3 N aq HCl. The solid was collected and dried to give 6.0 g (64%) of **3**: mp 347 °C dec: <sup>1</sup>H NMR (Me<sub>2</sub>SO-*d*<sub>6</sub>) δ 11.75 (br s, exchanges with D<sub>2</sub>O, 1H), 11.49 (br s, exchanges with D<sub>2</sub>O, 1H), 8.40 (s, 1H), 7.93 (s, 1H), 7.63 (s, 1H), 7.62 (dd, *J* = 3.0, 2.4 Hz), 6.68 (dd, *J* = 3.0, 0.9 Hz); <sup>13</sup>C NMR δ 161.7, 143.0, 142.2, 140.0, 129.4, 127.8, 117.6, 115.5, 107.2, 102.1; CIMS *m/z* (relative intensity) 186 (MH<sup>+</sup>, 100), 185 (61). Anal. Calcd for C<sub>10</sub>H<sub>7</sub>N<sub>3</sub>O: C, 64.86; H, 3.81; N, 22.69. Found: C, 64.57; H, 3.64; N, 22.66.

**Method B.** A suspension of 0.25 g (1.4 mmol) of indole amide **22** and 7 mL of triethyl orthoformate was heated at 100 °C for 1 h. The mixture was cooled to 25 °C and filtered, and the filtrate was concentrated to a solid that was triturated in hot EtOAc. The solids were collected and dried to leave 0.13 g (50%) of pure **3**, mp > 280 °C, identical with material obtained by method A.

**Method C.** A mixture of 0.6 g (1.8 mmol) of **20** and 0.2 g of Raney nickel in 50 mL of 1:1 THF:MeOH was hydrogenated in a rocking autoclave at 1500 psi at 25 °C for 22 h. The catalyst was filtered, and the filtrate was concentrated. The crude solid was triturated in *i*-PrOH, collected, washed with *i*-PrOH and Et<sub>2</sub>O, and then dried to leave 0.19 g (57%) of **3**, mp > 280 °C, shown to contain minor impurities by <sup>1</sup>H NMR.

**5-Methyl-2-nitrobenzoic Acid, Ethyl Ester (24).** A solution of 99 g (0.54 mol) of acid **23**, 1 L of EtOH, and 70 g of concd H<sub>2</sub>SO<sub>4</sub> was heated at reflux for 2 d. The solution was concentrated, and the resultant oil was diluted with 300 mL of EtOAc. The organic phase was washed with 1% aq NaOH until the aqueous phase remained basic, dried, and concentrated *in vacuo* at 40 °C overnight to afford 103 g (91%) of **24** as a yellow oil: <sup>1</sup>H NMR (Me<sub>2</sub>SO-*d*<sub>6</sub>) δ 7.89 (d, *J* = 8.2 Hz, 1H), 7.64 (s, 1H), 7.60 (d, *J* = 8.2 Hz, 1H), 4.31 (q, *J* = 7.0 Hz, 2H), 2.45 (s, 3H), 1.27 (t, *J* = 7.0 Hz, 3H); <sup>13</sup>C NMR (Me<sub>2</sub>SO-*d*<sub>6</sub>) δ 165.1, 145.0, 132.6, 129.8, 127.2, 124.2, 62.0, 20.7, 13.7; CIMS *m/z* (relative intensity) 210 (MH<sup>+</sup>, 3), 164 (100). Anal. Calcd for C<sub>10</sub>H<sub>11</sub>N<sub>2</sub>O<sub>4</sub>: C, 57.41; H, 5.30; N, 6.70. Found: C, 57.46; H, 5.34; N, 6.55.

**2,6-Dinitro-*m*-toluic Acid, Ethyl Ester (25).** To a solution of 9.7 g (64 mmol) of CF<sub>3</sub>SO<sub>3</sub>H in 20 mL of CH<sub>2</sub>Cl<sub>2</sub> was added

over 10 min 2.3 g (34 mmol) of fuming HNO<sub>3</sub>. The resulting suspension was stirred at 25 °C for 10 min, and then 7.1 g (34 mmol) of ester **24** was added portionwise. The mixture was stirred at 25 °C for 20 h and then poured carefully into stirring ice cold H<sub>2</sub>O. The aqueous phase was separated and further extracted with CH<sub>2</sub>Cl<sub>2</sub>. The combined organic extracts were washed with 1% aq NaOH until the aqueous phase remained basic and then with H<sub>2</sub>O, dried, and concentrated to a yellow oil. Column chromatography eluting with hexanes:EtOAc (8:1) afforded 2.7 g (31%) of **25** as a solid, mp 58–59 °C (lit.<sup>13</sup> mp 58–59 °C): <sup>1</sup>H NMR (Me<sub>2</sub>SO-*d*<sub>6</sub>) δ 8.30 (d, *J* = 8.5 Hz, 1H), 7.86 (d, *J* = 8.5 Hz, 1H), 4.29 (q, *J* = 7.1 Hz, 2H), 2.43 (s, 3H), 1.20 (t, *J* = 7.1 Hz, 3H); <sup>13</sup>C NMR (Me<sub>2</sub>SO-*d*<sub>6</sub>) δ 162.4, 148.2, 145.1, 138.7, 135.6, 127.9, 123.2, 63.8, 18.4, 13.8; CIMS *m/z* (relative intensity) 254 (M<sup>+</sup>, 4.6), 209 (100). Anal. Calcd for C<sub>10</sub>H<sub>10</sub>N<sub>2</sub>O<sub>6</sub>: C, 47.25; H, 3.97; N, 11.02. Found: C, 47.31; H, 3.98; N, 11.00. Further workup afforded 2.6 g (30%) of isomeric compound **14**.

**(*E*)-3-[2-(Dimethylamino)ethenyl]-2,6-dinitrobenzoic Acid, Ethyl Ester (26).** A mixture of 1.1 g (4.9 mmol) of ester **25**, 1.3 mL (9.7 mmol) of DMF dimethyl acetal, and 5 mL of DMF was heated at 100 °C for 4 h. The solution was cooled and poured into H<sub>2</sub>O, and the precipitated red solid was collected. The solid was washed with H<sub>2</sub>O and dried to afford 1.2 g (70%) of **26**: mp 174–175 °C: <sup>1</sup>H NMR (Me<sub>2</sub>SO-*d*<sub>6</sub>) δ 7.95 (d, *J* = 9.3 Hz, 1H), 7.87 (d, *J* = 12.7 Hz, 1H), 7.78 (d, *J* = 9.3 Hz, 1H), 4.92 (d, *J* = 12.7 Hz, 1H), 4.23 (q, *J* = 7.1 Hz, 2H), 2.90 (br s, 6H), 1.19 (t, *J* = 7.1 Hz, 3H); <sup>13</sup>C NMR (Me<sub>2</sub>SO-*d*<sub>6</sub>) δ 163.4, 151.4, 141.4, 141.3, 136.7, 126.7, 125.7, 123.2, 86.3, 63.1, 13.8; CIMS *m/z* (relative intensity) 310 (MH<sup>+</sup>, 81), 309 (32). Anal. Calcd for C<sub>13</sub>H<sub>15</sub>N<sub>3</sub>O<sub>6</sub>: C, 50.49; H, 4.89; N, 13.59. Found: C, 50.94; H, 4.93; N, 13.51.

**6-Amino-1H-indole-7-carboxylic Acid, Ethyl Ester (27).** A mixture of 1.1 g (3.5 mmol) of enamine **26**, 0.5 g of 10% Pd/C, and 100 mL of MeOH was hydrogenated at 50 psi for 4 h. The mixture was filtered, concentrated, and then purified by column chromatography eluting with 4:1 hexanes:EtOAc. The product fractions were concentrated to a solid that was triturated from hexanes to afford 0.35 g (48%) of **27**: mp 81–82 °C: <sup>1</sup>H NMR (Me<sub>2</sub>SO-*d*<sub>6</sub>) δ 10.3 (br s, exchanges with D<sub>2</sub>O, 1H), 7.39 (d, *J* = 8.6 Hz, 1H), 6.95 (m, 1H), 6.63 (s, exchanges with D<sub>2</sub>O, 2H), 6.45 (dd, *J* = 8.7, 2.4 Hz, 1H), 6.23 (m, 1H), 4.36 (q, *J* = 7.3 Hz, 2H), 1.31 (t, *J* = 7.3 Hz, 3H); <sup>13</sup>C NMR (Me<sub>2</sub>SO-*d*<sub>6</sub>) δ 168.1, 149.9, 135.0, 127.8, 122.2, 119.0, 110.7, 102.4, 94.0, 60.0, 15.1; CIMS *m/z* (relative intensity) 204 (M<sup>+</sup>, 100). Anal. Calcd for C<sub>11</sub>H<sub>12</sub>N<sub>2</sub>O<sub>2</sub>: C, 64.69; H, 5.92; N, 13.72. Found: C, 64.66; H, 5.98; N, 13.64.

**1,6-Dihydro-9H-pyrrolo[2,3-*f*]quinazolin-9-one (4).** A solution of 0.10 g (0.49 mmol) of indole **17** in 2 mL of formamide was heated for 1 h each at 130 °C, 250 °C, and 180 °C. The mixture was cooled and poured into ice-cold H<sub>2</sub>O. The aqueous suspension was extracted with EtOAc, and the organic layer was washed with brine, dried, and concentrated. Purification of the oily residue by column chromatography eluting with 9:1 CH<sub>2</sub>-Cl<sub>2</sub>:MeOH gave a solid that was crystallized from *ca.* 2:1 EtOAc:hexanes to provide 0.042 g (44%) of **4**: mp 297–298 °C; <sup>1</sup>H NMR (Me<sub>2</sub>SO-*d*<sub>6</sub>) δ 12.3 (br s, exchanges with D<sub>2</sub>O, 1H), 11.5 (br s, exchanges with D<sub>2</sub>O, 1H), 8.03 (s, 1H), 7.95 (d, *J* = 8.3 Hz, 1H), 7.38 (dd, *J* = 3.0, 2.6 Hz, 1H), 7.27 (d, *J* = 8.3 Hz, 1H), 6.59 (dd, *J* = 3.0, 2.2 Hz, 1H); <sup>13</sup>C NMR (Me<sub>2</sub>SO-*d*<sub>6</sub>) δ 160.8, 145.5, 142.9, 130.8, 127.3, 126.7, 125.6, 118.0, 107.9, 102.1; CIMS *m/z* (relative intensity) 186 (MH<sup>+</sup>, 96), 185 (100). Anal. Calcd for C<sub>10</sub>H<sub>7</sub>N<sub>3</sub>O: C, 64.86; H, 3.81; N, 22.69. Found: C, 64.59; H, 3.79; N, 22.59.

JO951790B